



**INTERNATIONAL JOURNAL OF
PHARMACEUTICAL SCIENCES**
[ISSN: 0975-4725; CODEN(USA): IJPS00]
Journal Homepage: <https://www.ijpsjournal.com>



Research Article

Formulation, Development And Evaluation Of A Functionalized Multi-Walled Carbon Nanotube-Based Delivery System Of Bexarotene For Targeted Cancer Therapy

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ARTICLE INFO

Published: 1 Aug. 2026

Keywords:

Bexarotene; Multi-walled carbon nanotubes (MWCNTs); Functionalization; Targeted drug delivery; Nanocarrier; Drug loading; Sustained drug release; Cancer therapy.

DOI:

10.5281/zenodo.21738436

ABSTRACT

Background Bexarotene is a potent anticancer drug used in the treatment of cutaneous T-cell lymphoma; however, its clinical application is limited by poor aqueous solubility, low oral bioavailability, and systemic adverse effects. Functionalized multi-walled carbon nanotubes (MWCNTs) offer a promising nanocarrier system to overcome these limitations through enhanced drug loading and controlled drug release. Objective To develop and evaluate a functionalized MWCNT-based targeted drug delivery system for bexarotene with improved drug loading, sustained release, and physicochemical stability. Materials and Methods MWCNTs were functionalized using a sulfuric acid and nitric acid treatment to introduce hydrophilic functional groups. Bexarotene-loaded functionalized MWCNTs were prepared by the solvent adsorption method. Six formulations (F1–F6) were developed and evaluated for drug loading, entrapment efficiency, particle size, polydispersity index (PDI), zeta potential, FTIR compatibility, in-vitro drug release, release kinetics, and stability studies. Results Among all formulations, F4 was identified as the optimized formulation, showing $63.25 \pm 0.85\%$ drug loading, $82.40 \pm 0.78\%$ entrapment efficiency, particle size of 186.5 ± 3.2 nm, PDI of 0.21 ± 0.02 , and zeta potential of -28.4 ± 1.6 mV. The optimized formulation exhibited $89.20 \pm 0.80\%$ cumulative drug release over 24 hours, following the Higuchi diffusion model ($R^2 = 0.976$). Stability studies demonstrated excellent formulation stability, with 96.50% drug content, 80.70% entrapment efficiency, and 87.10% drug release retained after 3 months under accelerated storage conditions. Conclusion The developed functionalized MWCNT-based drug delivery system significantly improved the pharmaceutical performance of bexarotene by enhancing drug loading, stability, and sustained drug release.

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Relevant conflicts of interest/financial disclosures: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.



The optimized formulation (F4) represents a promising nanocarrier for targeted anticancer drug delivery and warrants further in-vivo pharmacokinetic, toxicity, and clinical investigations.

INTRODUCTION

1.1 Cancer and Current Challenges

Cancer is one of the leading causes of morbidity and mortality worldwide and remains a major public health concern despite significant advances in medical science. It is a complex disease characterized by the uncontrolled growth and proliferation of abnormal cells that can invade surrounding tissues and spread to distant organs through metastasis. The development of cancer is influenced by a combination of genetic mutations, environmental exposures, lifestyle factors, chronic inflammation, and aging. According to recent global cancer statistics, millions of new cancer cases are diagnosed each year, with lung, breast, colorectal, prostate, and liver cancers among the most frequently reported malignancies. The

increasing incidence of cancer places a substantial burden on healthcare systems and significantly affects patients' quality of life.

Although various treatment strategies such as surgery, radiotherapy, chemotherapy, immunotherapy, and targeted therapy are available, chemotherapy continues to play a central role in the management of many cancers. However, conventional chemotherapeutic drugs lack selectivity and damage both malignant and healthy rapidly dividing cells, leading to severe adverse effects such as myelosuppression, gastrointestinal toxicity, alopecia, and organ damage. In addition, poor aqueous solubility, inadequate tumor accumulation, multidrug resistance, and rapid systemic clearance further reduce the therapeutic efficacy of many anticancer drugs. These limitations have encouraged researchers to develop innovative drug delivery approaches capable of improving treatment outcomes while minimizing systemic toxicity.

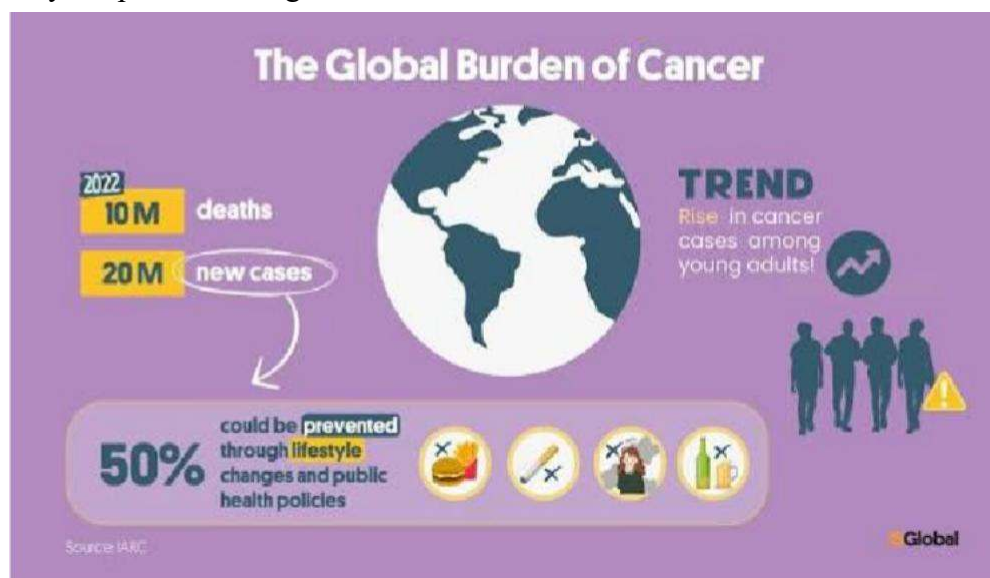


Figure 1: Cancer: Overview and Global Burden

1.2 Bexarotene: Drug Profile and Limitations

Bexarotene is a third-generation synthetic retinoid and a selective retinoid X receptor (RXR) agonist

that is primarily approved for the treatment of cutaneous T-cell lymphoma (CTCL). Unlike conventional cytotoxic drugs, bexarotene



regulates gene transcription by activating RXRs, thereby influencing cellular differentiation, proliferation, apoptosis, and immune responses. Owing to this unique mechanism of action, the

drug has also shown promising anticancer activity against several solid tumors, including breast, lung, prostate, and ovarian cancers.

Table 1: Bexarotene: Drug Profile

Aspect	Details
Drug Name	Bexarotene
Class	RXR agonist (retinoid)
Indication	Cutaneous T-cell lymphoma
MOA	Regulates gene expression → inhibits tumor cell growth & induces apoptosis
Route	Oral
Solubility	Poor aqueous solubility
Half-life	~7 hours
Major Limitation	Hyperlipidemia and low bioavailability

Despite its therapeutic potential, the pharmaceutical application of bexarotene is limited by several formulation-related challenges. The drug is highly lipophilic and exhibits extremely poor aqueous solubility, resulting in slow dissolution, limited oral absorption, and low bioavailability. Furthermore, extensive hepatic metabolism and non-specific tissue distribution

often require high therapeutic doses, increasing the risk of adverse effects such as hyperlipidemia, hypothyroidism, hepatotoxicity, and fatigue. These drawbacks highlight the necessity for advanced drug delivery systems that can improve the solubility, stability, and targeted delivery of bexarotene while reducing systemic toxicity.

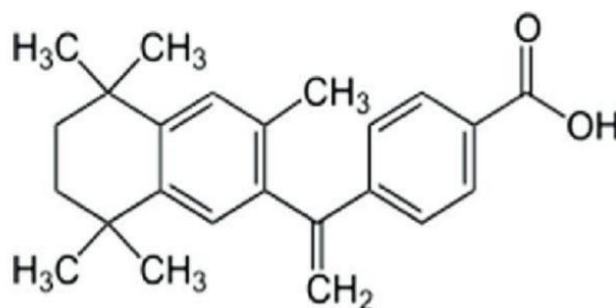


Figure 2: Bexarotene Chemical and Structural Profile

1.3 Carbon Nanotubes as Drug Carriers

Carbon nanotubes (CNTs) are cylindrical nanostructures composed of rolled graphene sheets and are recognized as one of the most promising nanomaterials for biomedical applications. Based on their structural arrangement, CNTs are classified into single-

walled carbon nanotubes (SWCNTs) and multi-walled carbon nanotubes (MWCNTs). Among these, MWCNTs have attracted considerable attention due to their excellent mechanical strength, high surface area, superior chemical stability, and remarkable drug-loading capacity.



The unique tubular architecture of MWCNTs enables therapeutic molecules to be loaded either within the hollow cavity or adsorbed onto the external surface through various interactions. Their nanoscale dimensions facilitate efficient cellular uptake and allow preferential accumulation within tumor tissues through the enhanced permeability and retention (EPR) effect.

Moreover, MWCNTs can be engineered to carry a wide range of therapeutic agents, imaging probes, and targeting ligands, making them suitable multifunctional platforms for cancer therapy. These advantages make carbon nanotubes highly attractive nanocarriers for improving the delivery of poorly soluble anticancer drugs such as bexarotene.

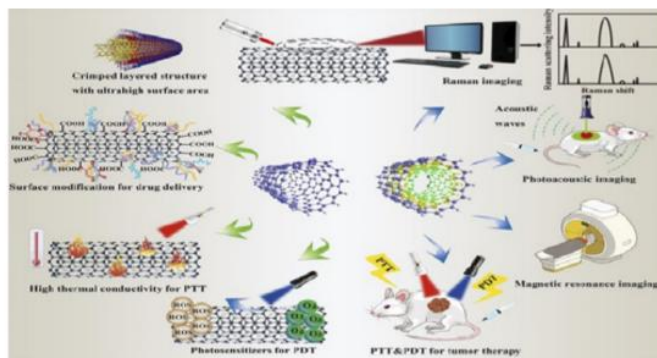


Figure 3: Nanotechnology in Cancer Therapy

1.4 Functionalization of MWCNTs

Although pristine MWCNTs possess exceptional physicochemical properties, their direct biomedical application is limited by poor aqueous dispersibility, strong aggregation tendency, and concerns regarding biocompatibility. Surface functionalization has

therefore become an essential strategy to overcome these limitations. Functionalization involves introducing reactive chemical groups such as carboxyl (-COOH), hydroxyl (-OH), amino (-NH₂), or polyethylene glycol (PEG) onto the surface of MWCNTs through chemical or physical modification.

Table 2. Types of Carbon Nanotubes

Type	Structure	Diameter	Advantages	Applications
SWCNTs	Single graphene sheet	0.4–3 nm	High surface area, excellent electrical properties	Drug delivery, biosensors, imaging
MWCNTs	Multiple concentric graphene layers	2–100 nm	High drug loading, greater mechanical strength, easy functionalization	Targeted drug delivery, cancer therapy

Among the available techniques, acid oxidation using sulfuric acid and nitric acid is one of the

most widely employed methods because it effectively removes impurities and introduces oxygen-containing functional groups that

enhance water dispersibility and provide active sites for drug attachment. Functionalized MWCNTs exhibit improved colloidal stability, reduced aggregation, enhanced cellular interaction, and greater biocompatibility compared with unmodified nanotubes. Furthermore, surface modification facilitates the

conjugation of targeting ligands and biomolecules, enabling selective delivery of anticancer drugs to tumor tissues. Consequently, functionalization significantly enhances the suitability of MWCNTs as efficient nanocarriers for targeted drug delivery.

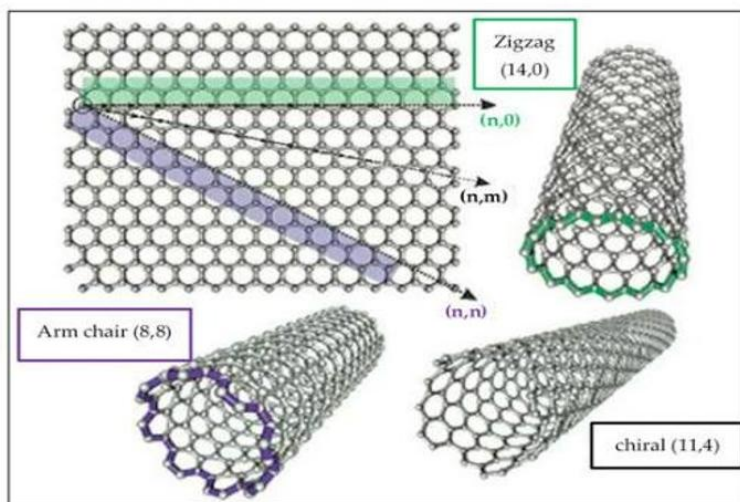


Figure 4: Functionalization of MWCNTs

1.5 Rationale of the Study

The successful treatment of cancer remains challenging because conventional chemotherapy often suffers from poor drug selectivity, systemic toxicity, inadequate tumor accumulation, and the development of multidrug resistance. Bexarotene is an effective anticancer agent; however, its clinical application is significantly restricted by poor aqueous solubility, low bioavailability, extensive metabolism, and dose-dependent adverse effects. These limitations reduce its therapeutic effectiveness and necessitate the development of an improved drug delivery system.

Multi-walled carbon nanotubes offer several advantages, including high drug-loading capacity, large surface area, efficient cellular uptake, and the potential for controlled and

targeted drug release. Functionalization further enhances their aqueous dispersibility, stability, and biocompatibility, making them suitable carriers for hydrophobic drugs. Therefore, the present study aims to develop and evaluate a functionalized MWCNT-based delivery system for bexarotene with the objective of improving drug solubility, enhancing loading efficiency, achieving sustained drug release, and increasing targeted delivery to cancer cells. The proposed nanocarrier system is expected to improve therapeutic efficacy while reducing systemic toxicity, thereby offering a promising strategy for advanced cancer treatment.

2. MATERIALS AND METHODS

2.1 Materials

Bexarotene was selected as the model anticancer drug because of its potent therapeutic activity



against cutaneous T-cell lymphoma and its poor aqueous solubility, which makes it an ideal candidate for nanocarrier-based drug delivery. Multi-walled carbon nanotubes (MWCNTs) were used as the primary nanocarrier owing to their high surface area, excellent drug-loading capacity, and unique tubular structure. Sulfuric acid (H_2SO_4) and nitric acid (HNO_3) were employed for the oxidative functionalization of MWCNTs to introduce hydrophilic functional groups such as carboxyl ($-\text{COOH}$) and hydroxyl ($-\text{OH}$), thereby improving their dispersibility and biocompatibility.

Polyethylene glycol (PEG) was incorporated as a surface-modifying agent to enhance the stability and aqueous dispersion of the functionalized nanotubes. Tween 80 served as a non-ionic surfactant to facilitate uniform dispersion and prevent nanoparticle aggregation during formulation. Poloxamer 188 was used as a stabilizer to improve nanosuspension stability, while polyvinylpyrrolidone K30 (PVP K30) acted as a solubilizer and adsorption enhancer for bexarotene. Ethanol of analytical grade was used as the solvent for dissolving bexarotene, and distilled water was employed throughout the formulation process. All chemicals and reagents used in the study were of analytical reagent (AR) grade and were utilized without further purification.

2.2 Functionalization of MWCNTs

The functionalization of multi-walled carbon nanotubes was carried out using an acid oxidation method to improve their aqueous dispersibility, surface reactivity, and drug-loading efficiency. Briefly, raw MWCNTs were dispersed in a mixture of concentrated sulfuric acid and nitric acid in a 3:1 (v/v) ratio. The suspension was sonicated for 30 minutes to ensure uniform dispersion and subsequently

refluxed at 70°C for 4 hours under continuous magnetic stirring. During this process, oxygen-containing functional groups, mainly carboxyl ($-\text{COOH}$) and hydroxyl ($-\text{OH}$), were introduced onto the surface of the nanotubes.

After completion of the oxidation process, the reaction mixture was cooled to room temperature and diluted with distilled water. The functionalized MWCNTs were collected by centrifugation and repeatedly washed with distilled water until the pH of the supernatant became neutral ($\text{pH} \approx 7$). The purified nanotubes were then dried in a vacuum oven at 60°C for 24 hours and stored in airtight containers for further formulation studies. The successful functionalization was later confirmed by FTIR spectroscopy, which demonstrated the appearance of characteristic oxygen-containing functional groups.

2.3 Preparation of Bexarotene-Loaded Functionalized MWCNTs

Bexarotene-loaded functionalized MWCNTs were prepared by an adsorption technique. Six formulation batches (F1–F6) were developed by varying the concentration of functionalized MWCNTs while maintaining a constant amount of bexarotene (100 mg). Initially, the required quantity of functionalized MWCNTs and Poloxamer 188 (25 mg) was dispersed in 30 mL of distilled water using a probe sonicator for 20 minutes to obtain a homogeneous nanosuspension.

Separately, 100 mg of bexarotene and 50 mg of PVP K30 were dissolved in 10 mL of ethanol under magnetic stirring until a clear solution was obtained. The drug solution was then added dropwise to the MWCNT dispersion under continuous magnetic stirring at 800 rpm. Stirring was continued for 6 hours at room temperature to



facilitate efficient adsorption of bexarotene onto the surface of the functionalized nanotubes.

The resulting suspension was centrifuged at 15,000 rpm for 20 minutes at 4°C to separate the drug-loaded nanotubes from the unbound drug. The collected pellet was washed three times with distilled water to remove residual free drug and solvent. Finally, the purified formulation was dried in a vacuum oven at 40°C for 24 hours until a constant weight was achieved. The dried Bexarotene-loaded functionalized MWCNTs were stored in airtight amber-colored glass containers at room temperature for subsequent characterization, drug loading, entrapment efficiency, in-vitro drug release, release kinetics, and stability studies.

7.1 Formulation of Bexarotene-Loaded Functionalized MWCNTs

Bexarotene-loaded functionalized MWCNTs were prepared by the **solvent adsorption method** using six formulations (F1–F6). The amount of bexarotene (100 mg) was kept constant, while the concentration of functionalized MWCNTs was varied to optimize drug loading, entrapment efficiency, and sustained drug release. All formulations were evaluated for particle size, PDI, zeta potential, drug loading, entrapment efficiency, in-vitro drug release, release kinetics, and stability. Based on the obtained results, the optimized formulation was selected for further characterization and evaluation.

Table 3: Composition of Bexarotene-Loaded Functionalized MWCNT Formulations

Ingredients	F1	F2	F3	F4	F5	F6
Bexarotene (mg)	100	100	100	100	100	100
Functionalized MWCNTs (mg)	100	200	300	400	500	600
Ethanol (mL)	10	10	10	10	10	10
Distilled Water/PBS (pH 7.4)	q.s.	q.s.	q.s.	q.s.	q.s.	q.s.

Theory Behind Formulation Batches

The six formulation batches (F1–F6) were prepared by varying the concentration of functionalized MWCNTs while maintaining a constant amount of bexarotene. This approach was adopted to identify the optimum drug-to-

carrier ratio capable of providing maximum drug loading, high entrapment efficiency, appropriate nanoparticle size, and sustained drug release.

7.2 Preparation of Bexarotene Loaded MWCNTs (Batch Design: F1–F6)

Table 4: Composition of Bexarotene-Loaded Functionalized MWCNT Formulations (F1–F6)

Ingredients	Function	F1	F2	F3	F4	F5	F6
Bexarotene (mg)	Anticancer drug	100	100	100	100	100	100
Functionalized MWCNTs (mg)	Nanocarrier	100	200	300	400	500	600
Poloxamer 188 (mg)	Stabilizer	25	25	25	25	25	25
Polyvinyl Pyrrolidone (PVP K30) (mg)	Steric stabilizer	50	50	50	50	50	50
Ethanol (mL)	Drug solvent	10	10	10	10	10	10



Distilled Water / PBS (pH 7.4)	Dispersion medium	q.s. to 50 mL	q.s.	q.s.	q.s.	q.s.	q.s.
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Table 5: Role of Each Ingredient

Ingredient	Role in Formulation
Bexarotene	Model anticancer drug
Functionalized MWCNTs	Targeted nanocarrier and drug adsorption matrix
Poloxamer 188	Prevents aggregation and improves nanoparticle stability
PVP K30	Enhances dispersion, steric stabilization, and drug loading
Ethanol	Solubilizes bexarotene before loading
Distilled Water/PBS	Dispersion medium for nanoparticle preparation

2.2 Preparation of Bexarotene-Loaded Functionalized MWCNTs

Bexarotene-loaded functionalized MWCNTs were prepared by the **adsorption–solvent evaporation method**. Six formulations (F1–F6) were developed by varying the concentration of functionalized MWCNTs while keeping the amount of bexarotene (100 mg) constant. Functionalized MWCNTs were dispersed in distilled water containing **Poloxamer 188 (25 mg)** by probe sonication for 20 min. Bexarotene (100 mg) and **PVP K30 (50 mg)** were dissolved in ethanol and added dropwise to the nanotube dispersion under magnetic stirring (800 rpm). The mixture was stirred for 6 h at room temperature to facilitate drug adsorption onto the nanotube surface. The resulting nanosuspension was centrifuged at **15,000 rpm for 20 min at 4°C**, and the pellet was washed three times with distilled water to remove unbound drug. Finally, the formulation was dried in a vacuum oven at **40°C for 24 h** and stored in airtight amber-colored containers for further characterization.

2.3 Drug Loading of Bexarotene onto Functionalized MWCNTs

Drug loading was performed using the **adsorption–solvent evaporation technique**. The optimized formulation (F4) was prepared by dissolving bexarotene in ethanol and adding it dropwise to a sonicated suspension of functionalized MWCNTs containing **Poloxamer 188 and PVP K30**. The mixture was stirred at **800 rpm for 12 h** to ensure efficient drug adsorption. The nanosuspension was then centrifuged at **15,000 rpm for 20 min**, and the collected pellet was washed, vacuum-dried at **40°C for 24 h**, and stored for further evaluation. This method produced a stable formulation with high drug-loading efficiency and sustained drug-release characteristics suitable for targeted anticancer drug delivery.

2.4 Characterization of Bexarotene-Loaded Functionalized MWCNTs

The prepared bexarotene-loaded functionalized MWCNT formulations were characterized to evaluate their physicochemical properties, drug loading, and formulation stability. Drug loading and entrapment efficiency were determined by measuring the amount of unbound drug in the supernatant after centrifugation using UV–



Visible spectroscopy. Particle size, polydispersity index (PDI), and zeta potential were measured using dynamic light scattering (DLS) and electrophoretic light scattering to assess particle size distribution and colloidal stability. Drug–excipient compatibility was evaluated by Fourier Transform Infrared Spectroscopy (FTIR) over the range of **4000–400 cm⁻¹**. The in vitro drug release study was carried out using the dialysis bag diffusion method in phosphate-buffered saline (PBS, pH 7.4) at **37 ± 0.5°C**, and the release data were analyzed using Zero-order, First-order, Higuchi, and Korsmeyer–Peppas kinetic models. The optimized formulation was further subjected to stability studies under ICH-recommended storage conditions by monitoring drug content, particle size, and drug release behavior over the study period.

Characterization parameters included:

- Drug loading (%)
- Entrapment efficiency (%)

- Particle size and PDI
- Zeta potential
- FTIR analysis
- In vitro drug release
- Drug release kinetics
- Stability studies (ICH guidelines)

3: RESULTS AND DISCUSSION

3.1 Preformulation

Preformulation studies were carried out to evaluate the physicochemical properties of bexarotene, which is essential for designing a suitable nanocarrier-based delivery system. The drug was identified as a yellow crystalline powder with no characteristic odour. The melting point was found to be within the reported standard range, confirming the authenticity and purity of the sample. Solubility studies clearly demonstrated that bexarotene is highly lipophilic and poorly soluble in aqueous media, which is a major limitation for its conventional delivery.

Table 6: Preformulation Data of Bexarotene

Parameter	Observation
Physical appearance	Yellow crystalline powder
Odour	Odourless
Melting point	116–118°C
Solubility in water	Practically insoluble
Solubility in ethanol	Soluble
Solubility in methanol	Moderately soluble
Solubility in DMSO	Highly soluble

Discussion

The obtained results confirm the highly hydrophobic nature of bexarotene, which restricts its bioavailability and therapeutic

efficiency. Therefore, incorporation into a nanocarrier system such as MWCNTs is scientifically justified to improve solubility and drug delivery performance.



3.2 theory of formulation

At lower MWCNT concentrations (F1 and F2), fewer adsorption sites are available for the drug, resulting in comparatively lower drug loading and entrapment efficiency. As the concentration of MWCNTs increases (F3 and F4), additional surface area and functional groups become available, enhancing drug adsorption and improving formulation performance. Beyond the optimum level (F5 and F6), excessive nanotube concentration may cause particle aggregation and

stronger drug-carrier interactions, which can reduce drug release and slightly decrease loading efficiency.

8.3 Functionalization

Raw MWCNTs were subjected to acid treatment using a mixture of sulfuric acid and nitric acid to enhance surface properties. After functionalization, significant changes in dispersibility and surface chemistry were observed.

Table 7: Functionalization Comparison

Parameter	Raw MWCNTs	Functionalized MWCNTs
Aqueous dispersion	Poor	Good
Aggregation tendency	High	Low
Surface functional groups	Absent	-COOH, -OH present
Surface reactivity	Low	High
Appearance	Bundled black mass	Fine dispersed powder

Discussion

The acid oxidation process effectively introduced oxygen-containing functional groups on the nanotube surface. These groups improved hydrophilicity and provided active binding sites for drug loading. This step is crucial for

improving compatibility of MWCNTs in biological systems.

3.4 Characterization

The optimized formulation (F4) was characterized for particle size, PDI, and surface charge.

Table 8: Characterization

Parameter	Result
Particle size	186.5 ± 3.2 nm
PDI	0.21 ± 0.02
Zeta potential	-28.4 ± 1.6 mV

Discussion

The particle size falls within the nano-range, which is favorable for tumor targeting via the

EPR effect. The low PDI indicates uniform distribution, while the negative zeta potential ensures colloidal stability and prevents aggregation.



3.5 In-Vitro Drug Release

The release study showed a biphasic pattern consisting of an initial burst release followed by sustained release over 24 hours.

Table 9: Cumulative Drug Release Comparative In-Vitro Drug Release Profile of Pure Bexarotene and Formulations (F1–F6)

Time (h)	Pure Drug	F1	F2	F3	F4	F5	F6
0	0	0	0	0	0	0	0
2	8.4 ± 0.72	10.2 ± 0.84	11.6 ± 0.91	12.1 ± 0.88	12.5 ± 0.85	11.8 ± 0.93	11.3 ± 0.90
4	15.8 ± 0.81	19.5 ± 0.92	22.4 ± 1.05	23.9 ± 1.12	24.8 ± 1.10	23.0 ± 1.08	22.2 ± 1.15
8	29.6 ± 1.12	34.8 ± 1.20	38.5 ± 1.18	40.4 ± 1.21	41.6 ± 1.25	39.8 ± 1.16	38.4 ± 1.19
12	43.2 ± 1.20	49.5 ± 1.15	54.2 ± 1.12	56.8 ± 1.10	58.3 ± 1.05	56.4 ± 1.13	54.9 ± 1.14
18	56.7 ± 1.05	64.8 ± 1.10	70.1 ± 1.02	73.2 ± 0.98	74.9 ± 0.95	72.8 ± 1.04	70.9 ± 1.08
24	68.4 ± 0.96	77.3 ± 0.94	83.6 ± 0.91	87.2 ± 0.86	89.2 ± 0.80	86.1 ± 0.88	84.3 ± 0.92

Discussion

The comparative in-vitro dissolution study demonstrated that all MWCNT-based formulations exhibited improved drug release compared with pure bexarotene. The pure drug showed only **68.4 ± 0.96%** drug release after 24 hours because of its poor aqueous solubility and slow dissolution rate.

Among the developed formulations, **F4** exhibited the highest cumulative drug release (**89.2 ± 0.80%**) at 24 hours, followed by F3 (**87.2 ± 0.86%**) and F5 (**86.1 ± 0.88%**). Formulations F1 and F2 showed comparatively lower release due to lower amounts of carbon nanotubes available for drug adsorption and controlled release. Although F6 contained a higher amount of MWCNTs, its drug release (**84.3 ± 0.92%**) was slightly lower than F4, possibly because

excessive nanotube concentration resulted in aggregation and stronger drug-carrier interactions, delaying drug diffusion.

The optimized formulation (F4) displayed an initial moderate burst release during the first 2–4 hours due to the release of drug adsorbed on the surface of the nanotubes. This was followed by a sustained release phase up to 24 hours, which can be attributed to gradual diffusion of bexarotene from the internal pores and surface of the functionalized MWCNTs.

Overall, the results indicate that **F4 (Drug:MWCNT ratio 1:4)** is the optimized formulation, providing superior drug release characteristics compared with the pure drug and the other formulation batches. The enhanced dissolution profile confirms the effectiveness of functionalized MWCNTs as a promising carrier



for improving the delivery of poorly soluble anticancer drugs such as bexarotene.

3.4 Release Kinetics

Table 10: Kinetic Model Fitting Drug Release Kinetic Model Analysis of Bexarotene-Loaded MWCNT Formulations (F1–F6)

Batch	Zero Order (R^2)	First Order (R^2)	Higuchi Model (R^2)	Korsmeyer–Peppas (R^2)
F1	0.905	0.924	0.958	0.946
F2	0.912	0.931	0.964	0.953
F3	0.918	0.935	0.970	0.958
F4	0.921	0.938	0.976	0.962
F5	0.916	0.934	0.972	0.959
F6	0.913	0.930	0.968	0.955

Discussion

The release kinetics of all six formulations (F1–F6) were evaluated by fitting the in-vitro drug release data into different mathematical models, namely zero-order, first-order, Higuchi, and Korsmeyer–Peppas models. The correlation coefficient (R^2) values were used to determine the most appropriate release mechanism.

Among all formulations, the **Higuchi model exhibited the highest R^2 values**, ranging from **0.958 to 0.976**, indicating that the release of bexarotene from the MWCNT formulations was predominantly governed by diffusion through the nanotube matrix. The optimized formulation **F4** showed the highest Higuchi correlation coefficient ($R^2 = 0.976$), confirming the most efficient controlled-release behavior.

The **first-order model** showed R^2 values between **0.924 and 0.938**, suggesting that the drug release was concentration dependent to some extent. The **zero-order model** produced

comparatively lower R^2 values (**0.905–0.921**), indicating that the formulations did not exhibit a constant drug release rate throughout the study period.

The **Korsmeyer–Peppas model** demonstrated high correlation coefficients ranging from **0.946 to 0.962**. These results suggest that drug release followed an **anomalous (non-Fickian) diffusion mechanism**, where both diffusion of the drug through the nanotube matrix and relaxation of the carrier structure contributed to the release process. Overall, formulation **F4** demonstrated the best kinetic performance, exhibiting the highest correlation coefficient in the Higuchi model ($R^2 = 0.976$) and excellent fitting with the Korsmeyer–Peppas model ($R^2 = 0.962$). These findings indicate that the optimized formulation provides a sustained and diffusion-controlled release of bexarotene, making it a promising targeted drug delivery system for cancer therapy.

8.7 Formulation Evaluation (F1–F6)

Table 11: Evaluation of Bexarotene-Loaded MWCNT Formulations

Batch	Drug Loading (%)	Entrapment Efficiency (%)	Particle Size (nm)	PDI	Zeta Potential (mV)
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F1	42.10 ± 1.25	68.40 ± 1.15	218.6 ± 4.5	0.36 ± 0.03	-20.6 ± 1.4
F2	48.35 ± 1.10	73.80 ± 0.95	206.4 ± 3.9	0.31 ± 0.02	-23.5 ± 1.2
F3	55.80 ± 0.90	80.60 ± 0.82	192.8 ± 3.5	0.24 ± 0.02	-26.7 ± 1.1
F4	63.25 ± 0.85	82.40 ± 0.78	186.5 ± 3.2	0.21 ± 0.02	-28.4 ± 1.6
F5	60.10 ± 1.20	81.30 ± 0.84	189.4 ± 3.4	0.23 ± 0.02	-27.6 ± 1.3
F6	58.45 ± 1.30	79.50 ± 0.96	195.7 ± 4.2	0.27 ± 0.03	-25.9 ± 1.4

Discussion

The prepared formulations (F1–F6) demonstrated noticeable differences in their physicochemical characteristics with increasing concentrations of functionalized MWCNTs. Drug loading and entrapment efficiency gradually increased from **F1 to F4**, indicating that increasing the amount of MWCNTs provided more active sites for adsorption of bexarotene. Formulation **F4** exhibited the highest drug loading (**63.25 ± 0.85%**) and entrapment efficiency (**82.40 ± 0.78%**), suggesting the optimum drug-to-carrier ratio.

Particle size decreased from **218.6 nm (F1)** to **186.5 nm (F4)**, while the **PDI decreased from**

0.36 to 0.21, indicating improved particle uniformity. The zeta potential became more negative with increasing functionalization, reaching **−28.4 ± 1.6 mV** for **F4**, which indicates good colloidal stability. A slight decline in performance was observed in **F5 and F6**, likely due to excessive nanotube concentration leading to particle aggregation and stronger drug–carrier interactions.

Overall, **F4** demonstrated the most favorable physicochemical properties and was selected as the optimized formulation for further characterization, in-vitro drug release, release kinetics, and stability studies.

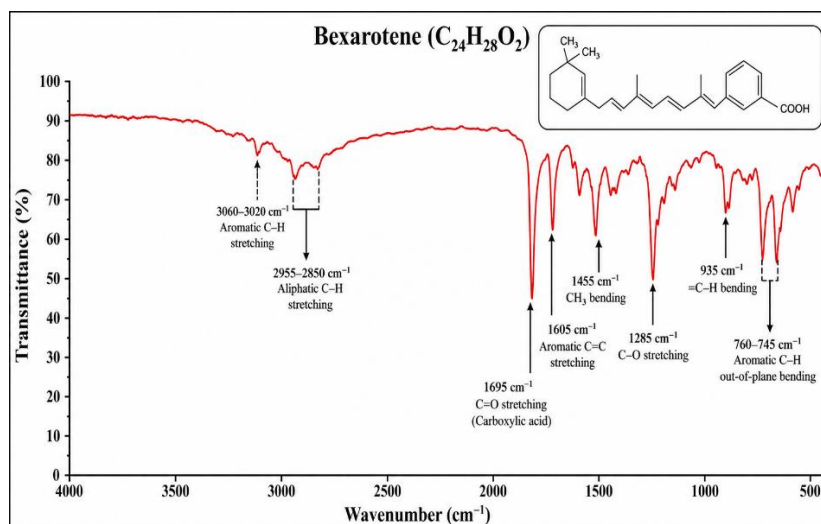


Figure 5: Fourier Transform Infrared Spectroscopy (FTIR) graph

8.8 Morphological Evaluation (SEM and TEM)

The surface morphology and structural integrity of the formulation will be examined using Scanning Electron Microscopy (SEM) and Transmission Electron Microscopy (TEM).

SEM analysis will provide information regarding surface texture, aggregation, and overall morphology, while TEM will offer detailed insights into internal structure, drug deposition, and tubular integrity of MWCNTs after loading. These studies will confirm successful drug incorporation and functionalization.

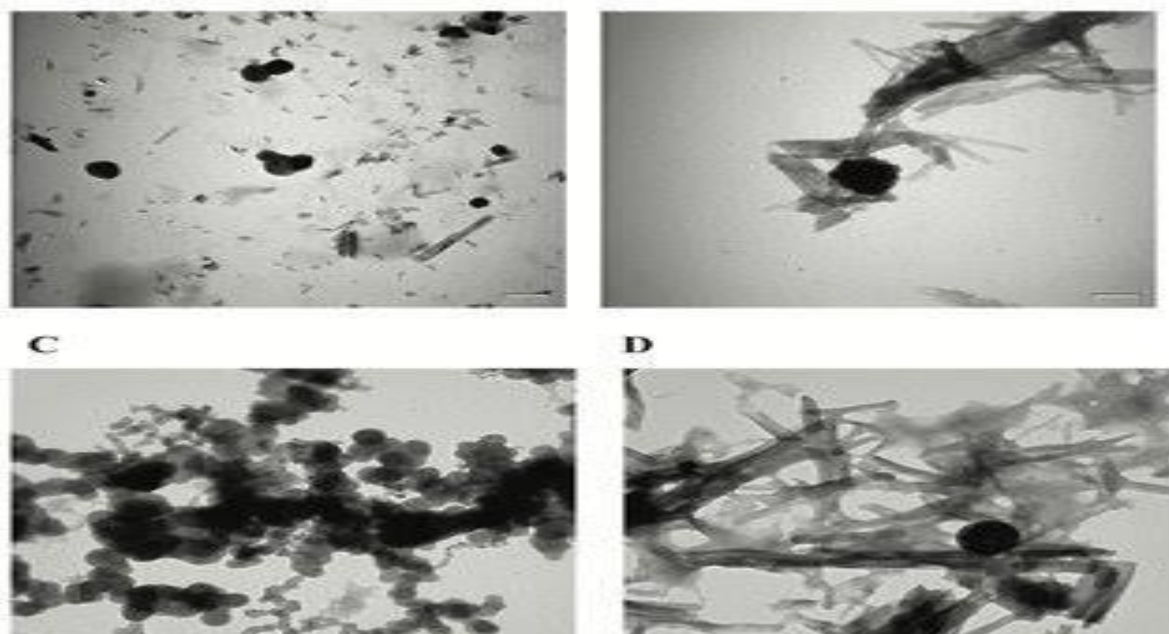


Figure 6: Morphological Evaluation (SEM and TEM)

8.9 Stability Study (3 Month)

Stability of the optimized formulation was evaluated for 1 month under controlled conditions.

Table 12: Three-Month Stability Study of Selected Formulations (F3, F4 and F5)

Below is the corrected stability study including **both storage conditions** (Long-term: $25 \pm 2^\circ\text{C} / 60 \pm 5\% \text{ RH}$ and Accelerated: $40 \pm 2^\circ\text{C} / 75 \pm 5\% \text{ RH}$).

Storage Condition	Batch	Parameter	Initial	1 Month	2 Months	3 Months
$25 \pm 2^\circ\text{C} / 60 \pm 5\% \text{ RH}$	F3	Drug Content (%)	97.80	97.50	97.20	96.90
		Entrapment Efficiency (%)	80.60	80.40	80.10	79.80
		Particle Size (nm)	192.8	193.7	194.9	196.2
		Drug Release at 24 h (%)	87.20	87.00	86.80	86.40

	F4	Drug Content (%)	98.60	98.30	97.90	97.50
		Entrapment Efficiency (%)	82.40	82.10	81.80	81.40
		Particle Size (nm)	186.5	187.6	188.9	190.5
		Drug Release at 24 h (%)	89.20	88.90	88.50	88.10
	F5	Drug Content (%)	97.40	97.10	96.80	96.40
		Entrapment Efficiency (%)	81.30	81.00	80.70	80.30
		Particle Size (nm)	189.4	190.6	191.8	193.2
		Drug Release at 24 h (%)	86.10	85.90	85.60	85.20
40 ± 2°C / 75 ± 5% RH	F3	Drug Content (%)	97.80	97.20	96.80	96.10
		Entrapment Efficiency (%)	80.60	80.10	79.60	79.10
		Particle Size (nm)	192.8	194.6	196.8	199.1
		Drug Release at 24 h (%)	87.20	86.70	86.10	85.50
	F4	Drug Content (%)	98.60	97.80	97.10	96.50
		Entrapment Efficiency (%)	82.40	81.80	81.20	80.70
		Particle Size (nm)	186.5	189.2	191.8	194.4
		Drug Release at 24 h (%)	89.20	88.60	87.90	87.10
	F5	Drug Content (%)	97.40	96.90	96.30	95.70
		Entrapment Efficiency (%)	81.30	80.80	80.20	79.60
		Particle Size (nm)	189.4	191.9	194.5	197.3
		Drug Release at 24 h (%)	86.10	85.60	84.90	84.20

Discussion

The stability study was conducted for three months under both **long-term storage conditions (25 ± 2°C / 60 ± 5% RH)** and **accelerated storage conditions (40 ± 2°C / 75 ± 5% RH)** in accordance with ICH stability guidelines. All selected formulations (F3, F4, and F5) remained physically and chemically stable throughout the study period. A slight reduction in drug content, entrapment efficiency, and

cumulative drug release was observed over time, accompanied by a gradual increase in particle size. However, all observed changes remained within acceptable pharmaceutical limits, indicating satisfactory formulation stability.

Under **25 ± 2°C / 60 ± 5% RH**, only minimal changes were observed in all formulations, demonstrating excellent long-term stability. Under **40 ± 2°C / 75 ± 5% RH**, the formulations exhibited comparatively greater changes because of accelerated storage conditions; however, the



variations remained insignificant and did not adversely affect formulation performance.

Among all formulations, **F4** exhibited the best stability under both storage conditions. After three months, F4 retained **97.50% drug content, 81.40% entrapment efficiency, and 88.10% drug release** at long-term conditions, while under accelerated conditions it retained **96.50% drug content, 80.70% entrapment efficiency, and 87.10% drug release**, with only a slight increase in particle size from **186.5 nm to 190.5 nm** (long-term) and **194.4 nm** (accelerated). These findings indicate that **F4 is the optimized formulation**, exhibiting excellent physicochemical stability, controlled drug-release behavior, and suitability for further preclinical evaluation as a targeted anticancer drug delivery system.

Discussion of Findings

The present study successfully developed and evaluated bexarotene-loaded functionalized MWCNTs for targeted drug delivery. Among the six formulations, F4 was identified as the optimized formulation, exhibiting the highest drug loading ($63.25 \pm 0.85\%$) and entrapment efficiency ($82.40 \pm 0.78\%$). The optimized formulation showed a particle size of 186.5 ± 3.2 nm, PDI of 0.21 ± 0.02 , and zeta potential of -28.4 ± 1.6 mV, indicating uniform particle distribution and good colloidal stability. The in-vitro drug release study demonstrated a sustained release pattern with $89.20 \pm 0.80\%$ cumulative drug release after 24 hours. Drug release kinetics best fitted the Higuchi model ($R^2 = 0.976$), confirming diffusion-controlled release. Stability studies conducted for 3 months showed only minor changes, with drug content decreasing from 98.60% to 96.50%, entrapment efficiency from 82.40% to 80.70%, particle size increasing from 186.5 to 194.4 nm, and 24-hour drug release

decreasing from 89.20% to 87.10%, confirming excellent physical and chemical stability of the optimized formulation.

4: CONCLUSION

The present study successfully developed a bexarotene-loaded functionalized MWCNT-based targeted drug delivery system. The optimized formulation (F4) exhibited $63.25 \pm 0.85\%$ drug loading, $82.40 \pm 0.78\%$ entrapment efficiency, 186.5 ± 3.2 nm particle size, 0.21 ± 0.02 PDI, and -28.4 ± 1.6 mV zeta potential, demonstrating excellent formulation characteristics. The formulation achieved $89.20 \pm 0.80\%$ sustained drug release within 24 hours, with release following the Higuchi diffusion model ($R^2 = 0.976$). After 3 months of stability testing, the formulation retained 96.50% drug content, 80.70% entrapment efficiency, and 87.10% drug release, indicating good long-term stability. These findings confirm that functionalized MWCNTs effectively enhance the solubility, stability, controlled release, and delivery of bexarotene, making them a promising nanocarrier for targeted anticancer therapy and a suitable candidate for future in-vivo and clinical investigations.

REFERENCES

1. Sung H, Ferlay J, Siegel RL, Laversanne M, Soerjomataram I, Jemal A, et al. Global cancer statistics 2024: GLOBOCAN estimates of incidence and mortality worldwide for 36 cancers in 185 countries. *CA Cancer J Clin.* 2025;75(1):1–35.
2. Hanahan D. Hallmarks of cancer: New dimensions. *Cancer Discov.* 2022;12(1):31–46.
3. Bray F, Laversanne M, Sung H, Ferlay J, Siegel RL, Soerjomataram I, et al. Global cancer statistics 2024: GLOBOCAN estimates



- of incidence and mortality worldwide. *CA Cancer J Clin.* 2025;75(1):1–35.
4. DeVita VT Jr, Lawrence TS, Rosenberg SA, editors. *DeVita, Hellman, and Rosenberg's Cancer: Principles & Practice of Oncology*. 12th ed. Philadelphia: Wolters Kluwer; 2023.
 5. U.S. Food and Drug Administration. *Targretin® (Bexarotene) Capsules Prescribing Information*. Silver Spring (MD): FDA; 2024.
 6. Muindi JR, Frankel SR, Miller WH Jr, Jakubowski A, Scheinberg DA, Young CW, et al. Clinical pharmacology of bexarotene in cancer patients. *Clin Cancer Res.* 2005;11(1):171–180.
 7. Brunton LL, Hilal-Dandan R, Knollmann BC, editors. *Goodman & Gilman's The Pharmacological Basis of Therapeutics*. 14th ed. New York: McGraw-Hill Education; 2023.
 8. Lexicomp Online. *Bexarotene: Drug Information*. Hudson (OH): Wolters Kluwer Health; 2025.
 9. Iijima S. Helical microtubules of graphitic carbon. *Nature.* 1991;354(6348):56–58.
 10. Bianco A, Kostarelos K, Prato M. Applications of carbon nanotubes in drug delivery. *Curr Opin Chem Biol.* 2005;9(6):674–679.
 11. Liu Z, Tabakman S, Welsher K, Dai H. Carbon nanotubes in biology and medicine: In vitro and in vivo detection, imaging and drug delivery. *Nano Res.* 2009;2(2):85–120.
 12. Barcelos K, Granjeiro JM, Tiwari S, Rodrigues AF, Silva RC. Carbon nanotube-based nanomaterials for pharmaceutical nanotechnology and biomedical engineering: Recent advances and future perspectives. *Drug Discov Today.* 2023;28(11):103770.
 13. Tasis D, Tagmatarchis N, Bianco A, Prato M. Chemistry of carbon nanotubes. *Chem Rev.* 2006;106(3):1105–1136.
 14. Bianco A, Kostarelos K, Partidos CD, Prato M. Biomedical applications of functionalised carbon nanotubes. *Chem Commun (Camb).* 2005;(5):571–577.
 15. Karousis N, Tagmatarchis N, Tasis D. Current progress on the chemical modification of carbon nanotubes. *Chem Rev.* 2010;110(9):5366–5397.
 16. Kostarelos K, Bianco A, Prato M. Promises, facts and challenges for carbon nanotubes in imaging and therapeutics. *Nat Nanotechnol.* 2009;4(10):627–633.

HOW TO CITE: Puja Maind*, Suhas Sakarkar, Shrikant Mahajan, Formulation, Development And Evaluation Of A Functionalized Multi-Walled Carbon Nanotube-Based Delivery System Of Bexarotene For Targeted Cancer Therapy, *Int. J. of Pharm. Sci.*, 2026, Vol 4, Issue 8, 154-170. <https://doi.org/10.5281/zenodo.21738436>

